

Generation of 6,7-Dihydro-5*H*-1,3,4-thiadiazolo[3,2-*a*]pyridines — Unexpected Products in the Reaction of 5-Ethylthio-1,3,4-thiadiazole-2-amine and 3-(Dimethylamino)propiophenones

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A series of 6,8-diaroyl-6,7-dihydro-5*H*-1,3,4-thiadiazolo[3,2-*a*]pyridines **4a–e** was prepared by the reaction of 5-ethylthio-1,3,4-thiadiazole-2-amine (**1**) and *p*-substituted 3-(dimethylamino)propiophenones **2a–e** in pyridine. The unexpected

cyclization process was established by NMR and X-ray diffraction measurements.

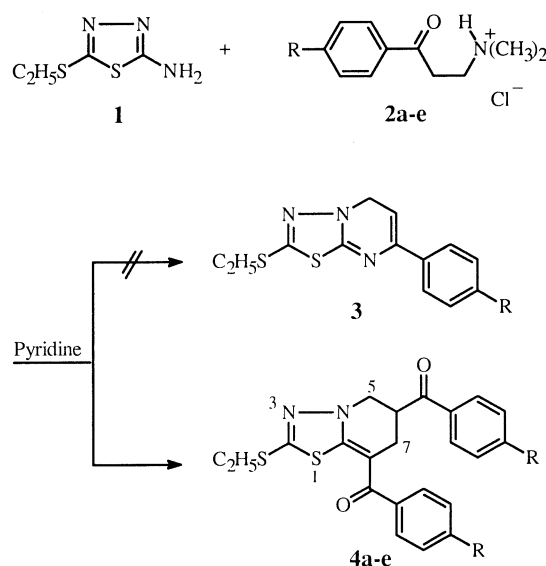
1,3,4-Thiadiazoles find a variety of applications as anti-biotic, antiinflammatory or bacteriostatic agents and as pesticides^{[1][2]}. These biological activities have stimulated us to synthesize and investigate a new class of fused 1,3,4-thiadiazoles. Based on our experience^{[3][4][5][6]} with the reaction

of aminoazoles and α,β -unsaturated ketones (chalcones) or their precursors, the 3-(dimethylamino)propiophenones, we studied the reaction of 5-ethylthio-1,3,4-thiadiazole-2-amine (**1**) with the hydrochlorides of *p*-substituted 3-(dimethylamino)propiophenones **2a–e** in boiling pyridine.

Although a condensation reaction of the amine **1** and the ketone followed by a cyclization to 5*H*-1,3,4-thiadiazolo[3,2-*a*]pyrimidines **3** seemed to be likely, we isolated 1:2-adducts which have the structure of 6,8-diaroyl-6,7-dihydro-5*H*-1,3,4-thiadiazolo[3,2-*a*]pyridines **4a–e**.

It is well-known that 3-aminopropiophenones are easily cleaved to aryl vinyl ketones which then can react with 3-aminopropiophenone in the sense of a Michael addition^[7].

Scheme 1



	R	m.p. [°C]	Yield [%]
4			
a	H	111	42
b	OCH ₃	136	45
c	Cl	155	50
d	Br	147	52
e	NO ₂	155	65

Scheme 2

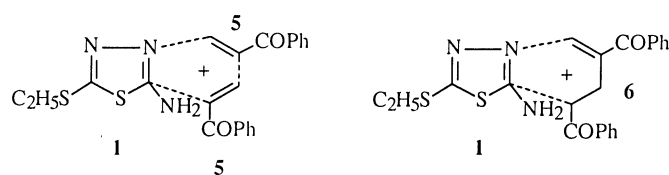


Table 1. ¹H-NMR data of the compounds **4a–e** (δ values in CD₃SOCD₃, TMS as the internal standard)

Comp.	5-H dd / dd	6-H m	7-H dd / dd	6-Aroyl m	8-Aroyl m	CH ₂ q	CH ₃ t
a	4.44 / 4.34	4.21	2.91 / 2.84	7.25–7.98	7.25–7.98	3.19	1.36
b	4.41 / 4.27	4.24	2.93 / 2.87	6.84–7.42	6.98–7.97	3.17	1.35
c	4.46 / 4.32	4.17	2.90 / 2.81	7.35–7.40	7.53–7.98	3.18	1.35 ^[a]
d	4.45 / 4.30	4.17	2.89 / 2.80	7.29–7.52	7.68–7.90	3.18	1.35
e	4.53 / 4.38	4.28	2.92 / 2.79	7.58–8.15	8.15–8.27	3.21	1.37

^[a] OCH₃ singlets at δ = 3.73 and 3.80

Table 2. ^{13}C -NMR data of the compounds **4a–e** (δ values in CD_3SOCD_3 , TMS as the internal standard)

4	C-2 / C-8a	C-5	C-6	C-7	C-8	6-CO	8-CO	CH_2	CH_3	CH	C_q
A	151.7 / 154.1	50.4	38.2	27.8	84.0	199.7	183.5	27.6	14.7	127.2, 127.9, 128.3, 128.8, 129.3, 133.5	135.1, 139.6
b	151.2 / 154.1	50.5	37.8	28.5	84.3	197.9	182.4	27.6	14.7 ^[a]	113.2, 114.0, 129.3, 130.7	127.9, 131.7, 160.2, 163.4
c	152.0 / 154.7	50.4	38.3	27.2	94.0	198.6	181.8	27.6	14.6	127.9, 128.9, 129.2, 130.2	133.7, 134.1, 138.1, 138.5
d	152.0 / 154.7	50.4	38.3	27.6	93.9	198.9	181.9	27.6	14.7	129.4, 130.3, 130.9, 131.9	122.8, 127.7, 134.0, 138.5
e	152.8 / 155.1	50.4	38.9	27.7	93.9	198.9	180.8	27.2	14.6	123.1, 123.8, 128.5, 129.7	139.7, 145.5, 147.5, 150.0

^[a] OCH_3 signals at $\delta = 51.5$ and 55.5

In the present case ketone **5** or its dimer **6** could be responsible for the anellation of a piperidine ring to the 1,3,4-thiadiazole ring. Extrusion of ammonia could then lead to the obtained compound **4**. However, it was not possible to find an experimental proof for the assumed intermediates under the reaction conditions used. Moreover, when we subjected equimolar amounts of **1** and **6** ^[7b] to the reaction conditions used, product **4** was not formed. Thus, the

mechanism for the unexpected process $\mathbf{1} + \mathbf{2} \rightarrow \mathbf{4}$ remains still unknown.

The structures of the compounds **4a–e** were established by ^1H - and ^{13}C -NMR measurements. Additionally, a crystal structure analysis was performed for **4b**. The tables 1 and 2 summarize the ^1H and ^{13}C chemical shifts. The assignment of the signals was supported by ^1H , ^1H COSY and ^1H , ^{13}C shift correlation spectra. The protons on the pyri-

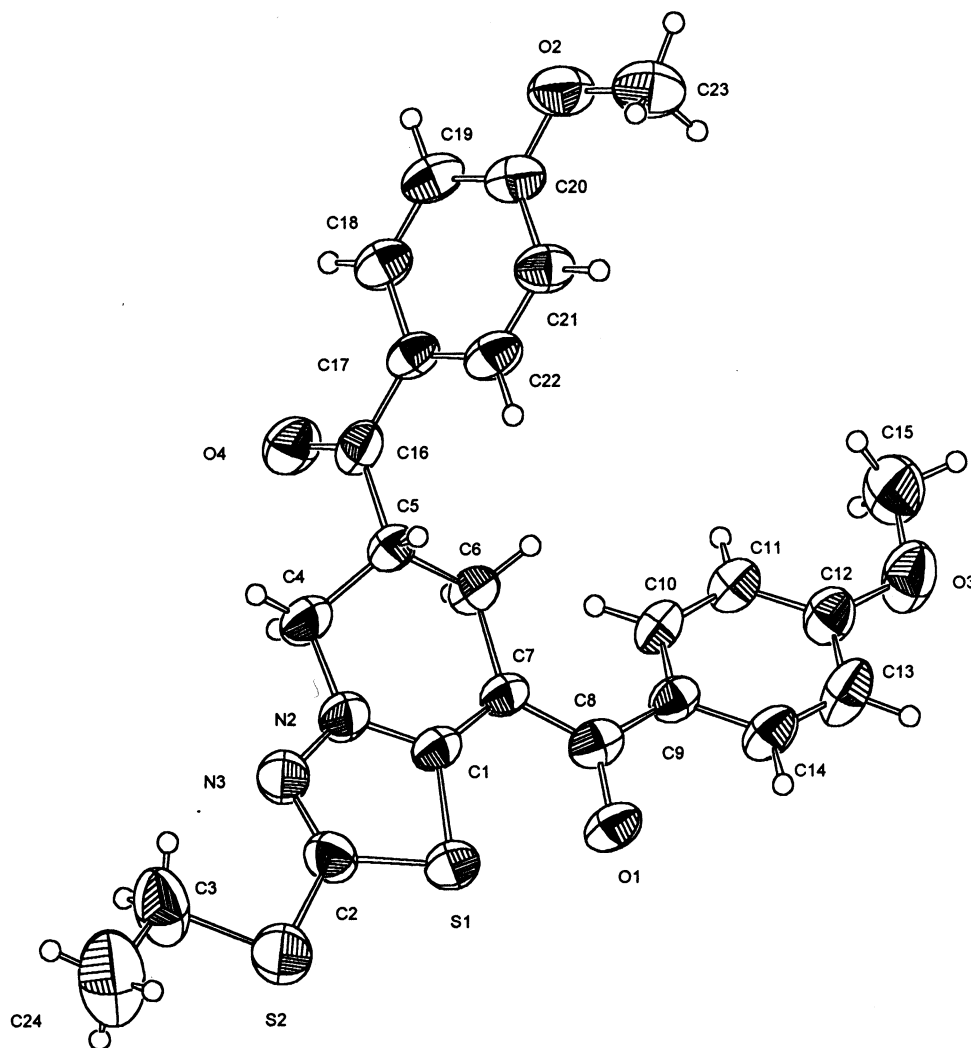
Figure 1. X-ray crystal structure of compound **4b** ^[11]

Table 3. Selected bond lengths [Å] bond angles [°] and torsional angles [°] of compound **4b** (The numbering corresponds to Figure 1 and not to the nomenclature)

S(1)–C(1)	1.738(3)	C(1)–S(1)–C(2)	88.1(2)
S(1)–C(2)	1.749(4)	N(3)–C(2)–S(1)	116.6(3)
S(2)–C(2)	1.743(3)	C(2)–N(3)–N(2)	108.3(3)
N(2)–C(1)	1.349(4)	C(1)–N(2)–N(3)	118.4(3)
N(2)–C(4)	1.464(4)	N(2)–C(1)–S(1)	108.6(2)
N(2)–N(3)	1.378(4)	N(2)–C(1)–C(7)	123.9(3)
N(3)–C(2)	1.288(4)	C(1)–C(7)–C(6)	117.0(3)
C(1)–C(7)	1.380(4)	C(7)–C(6)–C(5)	111.1(3)
C(6)–C(7)	1.516(4)	C(4)–C(5)–C(6)	110.8(3)
C(5)–C(6)	1.530(4)	N(2)–C(4)–C(5)	107.4(3)
C(4)–C(5)	1.514(4)	C(3)–S(2)–C(2)–N(3)	–2.5(4)
		C(4)–C(5)–C(16)–O(4)	–30.3(5)
		C(1)–C(7)–C(8)–O(1)	–5.0(5)
		C(4)–N(2)–C(1)–S(1)	–179.6(3)
		N(3)–N(2)–C(1)–C(7)	–179.6(3)

dine ring show a typical ABCDE pattern for the building block 5-CH₂–6-CH–7-CH₂. The geminal couplings ²*J* amount to (–12.5 ± 0.3) Hz for 5-CH₂ (AB) and to (–14.5 ± 0.3) Hz for 7-CH₂ (DE). The vicinal coupling constants are: ³*J*(AC) = 7.6 ± 0.3 Hz and ³*J*(CD) = 8.3 ± 0.3 Hz for the *trans* and ³*J*(BC) = 3.7 ± 0.2 Hz and ³*J*(CE) = 4.2 ± 0.2 Hz for the *cis* arranged protons. The integration of the four doublet of doublets (A, B, D, E) and the multiplet (C) corresponds to 1:1:1:1:1.

Figure 1 shows an X-ray crystal structure of compound **4b**; the most important geometric data of this novel type of heterobicyclic system^{[8][9][10]} are listed in table 3. The torsional angles reveal that the anellation of the two rings does not deviate much from planarity; the same statement can be made for the carbonyl group on C-8, which is cisoid and almost coplanar to the double bond between C-8 and C-8a.

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Experimental Section

Melting points were taken on a Büchi melting point apparatus and are uncorrected. The ¹H and ¹³C NMR spectra were run on a Bruker AM 400 in D₃CSOCD₃. The mass spectra were recorded on a Varian MAT 711 or a Finnigan M 95 operating at 70 eV. The elemental analyses were obtained by using a LEGO CHNS - 900 equipment.

General Procedure for the Preparation of the 6,7-Dihydro-5H-1,3,4-thiadiazolo[3,2-a]pyridines 4a–e: A solution of 5-ethylthio-1,3,4-thiadiazole-2-amine (**1**) (81 mg, 0.5 mmol) and 1.0 mmol of the corresponding 3-(dimethylamino)propiophenone hydrochloride (**2a–e**) in 4 ml of pyridine was heated for 20 min to reflux. The reaction products **4a–e** crystallized on cooling, were washed with cold ethanol and purified by column chromatography (2.5 x 40 cm silica gel) with petroleum ether (50–80 / ethyl acetate (1:1).

6,8-Dibenzoyl-2-ethylthio-6,7-dihydro-5H-1,3,4-thiadiazolo[3,2-a]pyridine (4a): Yellow crystals, m.p. 111°C, yield: 86 mg (42%). – MS, *m/z* (%): 408 (34) [M⁺], 304 (11), 303 (70), 105 (100)

[C₆H₅CO⁺]. – C₂₂H₂₀N₂O₂S₂ (408.6): calcd. C 64.68, H 4.93, N 6.86; found C 64.75, H 4.87, N 6.76.

2-Ethylthio-6,7-dihydro-6,8-bis(4-methoxybenzoyl)-5H-1,3,4-thiadiazolo[3,2-a]pyridine (4b): Yellow crystals, m.p. 136°C, yield: 105 mg (45%). – MS, *m/z* (%): 468 (32) [M⁺], 334 (17), 333 (86), 135 (100) [H₃CO–C₆H₄–CO⁺]. – C₂₄H₂₄N₂O₄S₂ (468.6): calcd. C 61.52, H 5.16, N 5.98; found C 61.65, H 5.07, N 5.86.

6,8-Bis(4-chlorobenzoyl)-2-ethylthio-6,7-dihydro-5H-1,3,4-thiadiazolo[3,2-a]pyridine (4c): Yellow crystals, m.p. 155°C, yield 119 mg (50%). – MS, *m/z* (%): 480 / 478 / 476 (4) [M⁺, Cl₂ pattern], 339 (13), 337 (31), 141 (31), 139 (100) [Cl–C₆H₄–CO⁺]. – C₂₂H₁₈Cl₂N₂O₂S₂ (477.5): calcd. C 55.34, H 3.80, N 4.95; found C 55.27, H 3.87, N 4.86.

6,8-Bis(4-bromobenzoyl)-2-ethylthio-6,7-dihydro-5H-1,3,4-thiadiazolo[3,2-a]pyridine (4d): Yellow crystals, m.p. 147°C, yield (147 mg) (52%). – MS, *m/z* (%): 568 / 566 / 564 (51) [M⁺, Br₂ pattern], 384 (17), 383 (100), 382 (18), 381 (93), 202 (10), 185 / 183 (22) [Br–C₆H₄–CO⁺]. – C₂₂H₁₈Br₂N₂O₂S₂ (566.4): calcd. C 46.66, H 3.20, N 4.95; found C 46.75, H 3.27, N 4.88.

2-Ethylthio-6,7-dihydro-6,8-bis(4-nitrobenzoyl)-5H-1,3,4-thiadiazolo[3,2-a]pyridine (4e): Pale orange crystals, m.p. 155°C, yield 162 mg (65%). – MS, *m/z* (%): 498 (18), [M⁺], 437 (15), 404 (36), 392 (16), 380 (18), 348 (39), 270 (11), 154 (11), 150 (59) [O₂N–C₆H₄–CO⁺], 122 (16), 120 (42), 104 (53), 92 (26), 87 (96), 64 (53), 59 (100). – C₂₂H₁₈N₄O₆S₂ (498.6): calcd. C 53.00, H 3.64, N 11.24; found C 53.05, H 3.57, N 10.99.

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- [11] C₂₄H₂₄N₂O₄S₂, molecular mass 468.70; crystal size 0.20 × 0.18 × 0.12 mm; triclinic, space group *P*-1; unit cell dimensions: *a* = 8.665(2) Å, *b* = 11.858(2) Å, *c* = 12.709(2) Å, *α* = 75.560(10)°, *β* = 70.000(10)°, *γ* = 71.210(10)°; *V* = 1147.4(4) Å³, *Z* = 2, *d* = 1.356 Mg/m³; *μ* = 0.266 mm^{–1}, *F*(000) = 492; *θ* range between 2.71 and 26.29°; index ranges 0 ≤ *h* ≤ 10, –13 ≤ *k* ≤ 14, –14 ≤ *l* ≤ 15; reflections collected 4940, independent reflections: 4625 [*R*(int) = 0.0367]; refinement: full-matrix least-squares on *F*²; goodness-of-fit on *F*²: 0.808; final *R* indices [*I* > 2 *σ* (*I*)] *R*1 = 0.0539, *ωR*2 = 0.0993; *R* indices (all data): *R*1 = 0.1627, *ωR*2 = 0.1170; extinction coefficient 0.0000(7); largest diff. peak and hole: 0.207 and –0.215 eÅ^{–3}. Further details of the structure investigation are available from the Fachinformationszentrum Karlsruhe, D-76344 Eggenstein-Leopoldshafen, Germany, on quoting the depository number CSD-406082 the names of the authors, and the journal citation.

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